ORGANIC CHEMISTRY



Total Marks: 36

Max. Time: 39 min.

Topic: Aldehydes, Ketones

	Type	of Q	uestions
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M.M., Min.

Single choice Objective ('-1' negative marking) Q.1 to Q.3

[12, 12] (3 marks 3 min.)

Comprehension ('-1' negative marking) Q.4 to Q.6

(3 marks 3 min.) [12, 12]

Subjective Questions ('-1' negative marking) Q.7

(4 marks 5 min.) [4, 5] (8 marks 10 min.) [8,10]

Match the Following (no negative marking) Q. 8

1.
$$CH_2(COOEt)_2 + OOEt$$

NaOEt

1. $dil. NaOH/\Delta \rightarrow \Delta \rightarrow COOLEt$

NaOEt

1. $dil. NaOH/\Delta \rightarrow \Delta \rightarrow COOLEt$

2. PhCHO + BrCH₂CO₂Et
$$\xrightarrow{\text{(1) Zn}}$$
 (A) $\xrightarrow{\text{(-H2O)}}$ (B)

Product of (B) of the above reaction is:

(A)
$$Ph - CH = CH - CH_3$$

(B)
$$Ph - CH = CH - CO$$
 Ft

(C)
$$Ph - CH = CH - CH_3 - OH$$

3.* The correct statement/s about the following reaction sequence is / are

- (A) 'R' gives an aldol condensation reaction on heating with NaOH solution
- (B) The compound 'Q' gives a yellow precipitate in acetone
- (C) Step '4' is an aromatic nucleophilic substitution reaction
- (D) The end product is a mixture of three compounds

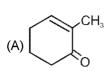
Comprehension # (Q. 4 to 6)

Intramolecular aldol condensation:

The aldol condensation also offer a convenient way to synthesize molecules with five and six membered ring. This can be done by an intramolecular aldol condensation using a dialdehyde, a keto aldehyde or a diketone as the substrate. The major product is formed by the attack of the enolate from the ketone side of the molecule that adds to the aldehyde group. The reason the aldehyde group undergoes addition preferentially may arise from the greater reactivity of aldehyde towards nucleophilic addition generally. In reaction of this type five membered rings form far more readily than seven membered rings and six membered rings are more favorable than four or eight membered rings when possible.



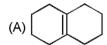
4. 1–Ethylcyclopent–1–ene on reductive ozonolysis followed by aq. NaOH/∆ gives

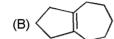


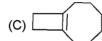


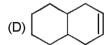
5. Which of the following compound on reaction with O₂/Zn, H₂O followed by aq. NaOH/∆ will form







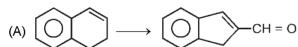




- - (A) It gives yellow ppt with I_{\star}/OH
- (B) It gives silver mirror with Tollen's reagent
- (C) It shows stereoisomerism
- (D) It does not give yellow ppt with 2,4 DNP
- 7. Match the correct reagents for the following transformations

Column-l (Transformations)

Column-II (Reagents)



(p) 1.
$$\frac{O_3}{Zn}$$
, 2. $\frac{NaOH}{\Delta}$

$$(B) \bigcirc \longrightarrow \bigcirc$$

(q) 1.
$$\frac{\mathrm{OsO_4}}{\mathrm{Na_2SO_3}}$$
 , 2. $\mathrm{HIO_4}$, 3. $\frac{\mathrm{KOH}}{\Delta}$

(C)
$$\bigcap_{N-CH_2-C-Ph}^{+CH_2-C-Ph} \longrightarrow \bigcap_{N-CH_2-C-Ph}^{O} \longrightarrow \bigcap_{N-C$$

(r) NaOH

(D)
$$Ph - C - C - Ph \longrightarrow Ph - C - COO OH$$

- (s) 1. $\frac{\text{HCN}}{\Delta}$, 2. $\text{H}_3\text{O}^{\oplus}$
- 8. How will you convert ethanal into the following compounds?
 - (i) Butane-1, 3-diol
- (ii) But-2-enal
- (iii) But-2-enoic acid



DPP No. # 27

2. (B)

CH3 - CH = CH - COOH

(B)

5.

7.
$$(A \rightarrow p; q); (B \rightarrow p, q); (C \rightarrow r); (D \rightarrow s)$$

8. (i)
$$CH_3$$
— $CH=O$ $\xrightarrow{\text{dil OH}^9}$ CH_3 $\xrightarrow{\text{CH}}$ CH_2 $\xrightarrow{\text{CH}}$ CH_2 $\xrightarrow{\text{CH}}$ CH_3 $\xrightarrow{\text{CH}}$ CH_3 $\xrightarrow{\text{CH}}$ CH_2 $\xrightarrow{\text{CH}}$ CH_3 $\xrightarrow{\text{CH}}$ CH_2 $\xrightarrow{\text{CH}}$ CH_3 $\xrightarrow{\text{CH}}$ CH_3 $\xrightarrow{\text{CH}}$ $CH=CH$ $CH=O$ $\xrightarrow{\text{OH}}$ CH_3 $\xrightarrow{\text{CH}}$ $CH=CH$ $CH=O$ $\xrightarrow{\text{OH}}$ CH_3 $\xrightarrow{\text{CH}}$ $CH=CH$ $CH=O$ $\xrightarrow{\text{CH}}$ CH_3 $\xrightarrow{\text{CH}}$ $CH=CH$ $CH=O$ $\xrightarrow{\text{CH}}$ CH_3 $CH=CH$ $CH=O$ $\xrightarrow{\text{CH}}$ $CH=CH$ $CH=O$ $CH=CH$ $CH=O$ CH

IS & 50

DPP No. #27

1.
$$CH_2(COOEt)_2 \xrightarrow{NaOEt} Na^{\bigoplus}CH^{\bigoplus}COOEt + \bigcirc COOEt + \bigcirc COOEt$$

3.*
$$\bigcirc$$
 CI^{\oplus} O_{2} $O_{$

$$CH = N.NH \longrightarrow NO_2 \longrightarrow (5) \longrightarrow (NHNH_2 NO_2 NO_2 (O)$$

4.
$$CH_{2}-CH_{3} \xrightarrow{O_{3}/Zn} CH_{2}-CH_{3} \xrightarrow{\mathring{O}H} CH_{2}-CH_{3} \xrightarrow{\mathring{O}H} CH_{3} \xrightarrow{(i)} H^{+} CH_{3} \xrightarrow{(i)} A$$

7. (A, B)
$$O_3/Zn \longrightarrow O_4/Zn \longrightarrow CH = O \longrightarrow CH = O \longrightarrow Aldol reaction$$

$$\begin{array}{c}
OSO_4 + Na_2SO_3 \\
OH
\end{array}$$

$$OH$$

$$OH$$

$$OHO_4$$

$$OHO$$

$$OHO$$

$$OHO$$
Aldol Reaction

(C) It is direct Aldol reaction

$$\begin{array}{c|c}
 & O \\
 & O \\$$

